

Thursday, Oct. 4th

- Check and go over homework from week
- Names and formulas for ionic compounds with polyatomic ions (next slide)
- Transition metals – Copper oxides
 - Transition Metal names and formulas
 - H/W practice

Name These Compounds

- $\text{Al}_2(\text{C}_2\text{O}_4)_3$
- $\text{Ba}(\text{NO}_3)_2$
- NH_4OH
- NaHCO_3

Write out the compound formula

- Ammonium carbonate
- Calcium phosphate
- Sodium hydroxide
- Potassium dichromate

Thursday Homework:

- Read pp 114-116
- Handout for bonding practice with transition metals – due Monday